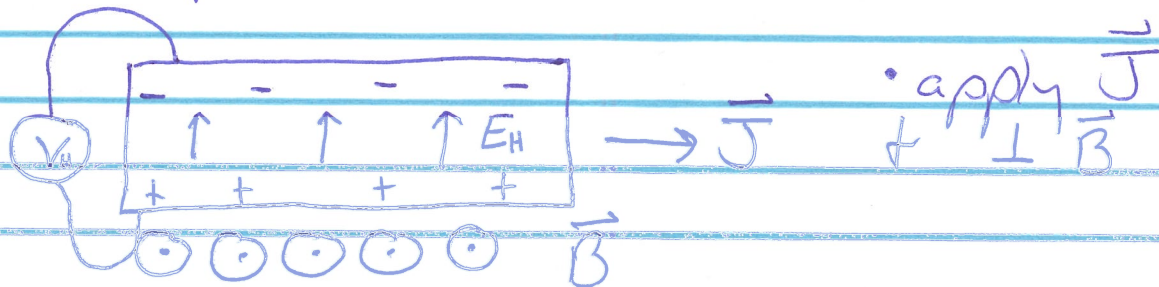


7/42

Wed. Sept. 12

①

- last time we discussed the Hall effect
- our example was a "slab" of metal



- measure V_H , the "Hall" voltage and convert to E_H

- we found $E_H = \left(\frac{-1}{ne} \right) JB = R_H JB$
Hall coeff. R_H

- We supply J + B + make measurement E_H

- $\Rightarrow n = \frac{-JB}{eE_H}$, so we may determine n

- or $n = \frac{-1}{eR_H} \Rightarrow \frac{n}{N_{\text{atom}}} = \frac{-1}{eR_H N_{\text{atom}}} = \text{ratio gives charge carriers per atom}$

$N_{\text{atom}} = \# \text{ density of atoms per volume}$

→

Valency + n/n_{atom} should match:

(2)

material	valence	n/n_{atom} (measured)
Li	1	0.8
Na	1	1.2
K	1	1.1
Ca	2	1.5
Mg	2	-0.4

- the model does a fair job for valency 1, but valency 2 sometimes gives weird answers (like neg. values)
- we will find out later this tells us something about the charge carriers for metals - like Mg (+) holes carry charge
- Recall that for Ohm's law, we could determine Σn , for scattering time constant τ
- now we can solve for $\tau \approx 10^{-14}$ s for most metals at room T.
- one last note about the Hall effect: we can convert voltage to B-field as a Hall probe
$$B = \frac{-neE_H}{J} \text{ (field sensor)}$$

③

- let's think about a 1D metal

$$\boxed{T_2 \quad T_1} \rightarrow \hat{x}$$

- the left side is hotter than the right $T_2 > T_1$

- heat crossing unit area per time

heat flux (W/m^2)

$$Q = -K \frac{dT}{dx}, \quad K \equiv \text{thermal conductivity}$$

- in metals, heat is conducted primarily by electrons since they are free to move
- if we again borrow from Kinetic Theory,

κ appa $\leftarrow$

$$K = \frac{1}{3} n C_v \langle v \rangle \lambda, \quad \lambda = \langle v \rangle \tau$$

(W/mK) scattering length

- for a monatomic gas $C_v = \frac{3}{2} k_B$

$$\langle v \rangle = \sqrt{\frac{8k_B T}{\pi m}} \rightarrow$$

- average speed for a gas at T

(4)

- which gives us

$$K = \frac{4}{\pi} \frac{n \tau k_B^2 T}{m}$$

- we determined τ from Ohm's law
- if we look at the ratio of thermal to electrical conductivity

$$\sigma = \frac{e^2 \tau n}{m} \quad \text{+ } K \text{ from above}$$

$$\bullet \quad \frac{K}{T\sigma} = \frac{4}{\pi} \left(\frac{k_B}{e} \right)^2 \sim 10^{-8} \frac{\text{Watt Ohm}}{\text{K}^2}$$

(very close to measured values)

$$\bullet \quad \text{Lorentz \# } \frac{1}{2}$$

- another way to look at $\frac{K}{\sigma}$

$$\frac{K}{\sigma} = \frac{1}{2} T \quad (\text{constant at fixed } T)$$

- matches experimental data for metals very well, but is wrong
- we don't get $C_v = \frac{3}{2} k_B$ for an electron

- specific heat $\sim 100\times$ too large
 - but $\langle v \rangle \sim 100\times$ too small
- } cancel giving artificial correct answer

(5)

- if the electrons carry heat from the hot end to the cold end, then the gradient gives rise to an electrical current
- the Peltier effect exploits this as applying a potential to a thermoelectric material makes a refrigerator
- conversely, a thermocouple turns a temperature gradient into a voltage $\rightarrow$ thus you have a thermometer (inexpensive & common)
- in 1925, Pauli discovered his "exclusion principle" that no two electrons may be in the exact same state
- Using this idea, we can describe the states of electrons in a conductor using Fermi-Dirac stats.
- Drude model assumed e^- 's can be in any state (classical) $\rightarrow$

(6)

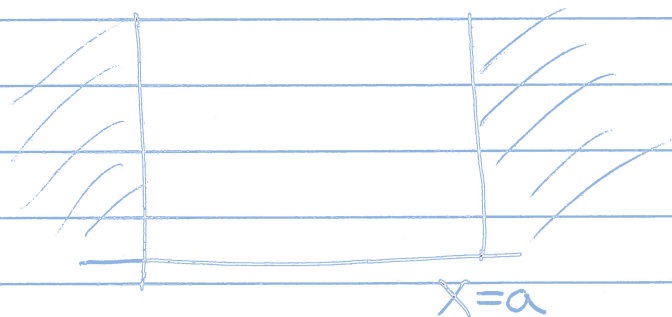
- let's do a little review of QM before moving into this discussion

- our time-ind. Schrödinger eq. in 1D

$$-\frac{\hbar^2}{2m} \frac{d^2\psi}{dx^2} + V\psi = E\psi$$

- recall the "particle in a box"

$$V(x) = \begin{cases} 0 & ; \quad 0 \leq x \leq a \\ \infty & ; \quad \text{otherwise} \end{cases}$$



- inside "box", $-\frac{\hbar^2}{2m} \frac{d^2\psi}{dx^2} = E\psi$

$$\frac{d^2\psi}{dx^2} = -k^2\psi, \quad k \equiv \frac{\sqrt{2mE}}{\hbar}$$

$$\psi(x) = A \sin(kx) + B \cos(kx)$$

(7)

- boundary conditions:

$$\psi(0) = \psi(a) = 0$$

$$\psi(0) = A \sin(0) + B \cos(0) = 0$$

$$B = 0$$

$$\psi(a) = A \sin(ka) = 0$$

$$ka = 0, \pm\pi, \pm 2\pi, \dots$$

$$k_n = \frac{n\pi}{a} = \frac{\sqrt{2mE_n}}{\hbar}$$

$$E_n = \frac{k_n^2 \hbar^2}{2m} \Rightarrow \boxed{E_n = \frac{n^2 \pi^2 \hbar^2}{2ma^2}}$$

- normal modes with energy E_n
- after normalizing

$$\psi_n(x) = \sqrt{\frac{2}{a}} \sin\left(\frac{n\pi}{a}x\right)$$

roughly, two metal atoms bond due to their free electrons having a lower energy in a larger "box" as $E_n \propto 1/a^2$